

Challenge #3: On the Cutting-edge

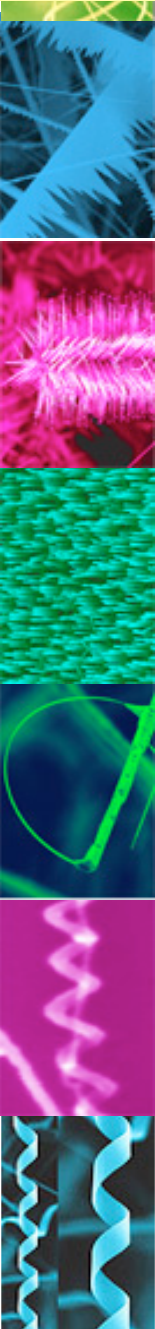
How many cuts are needed to reach 1nm?

$$10^{-7} \text{ cm} = 28 \text{ cm} \times 1/2^x$$

Need to solve for x: $1/2^x = 10^{-7}/28$

You can solve this in 2 ways:

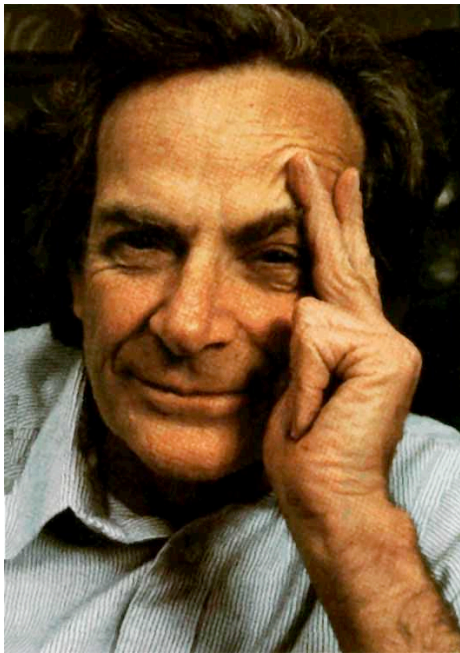
- exactly (using log and calculator) $x = 28$
- or by approximations ($28 = 32 = 2^5$ and $5 = 4 = 2^2$
 $\rightarrow 10^7 = 2^{21}$ or $5 = 8 = 2^3 \rightarrow 10^7 = 2^{28}$) $x = 26$; $x = 33$



How did we get here?

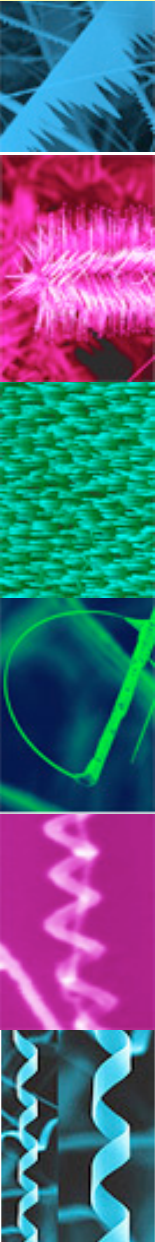
Origin of Nanoscience

- New Tools! As tools change, what we can see and do changes.



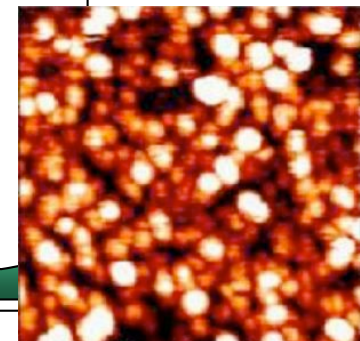
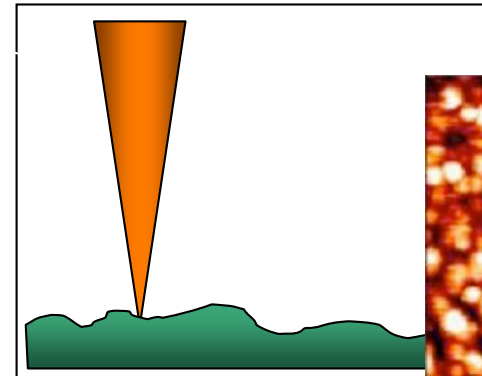
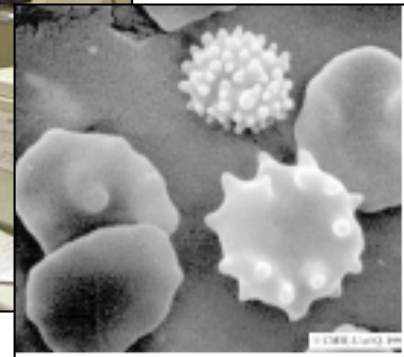
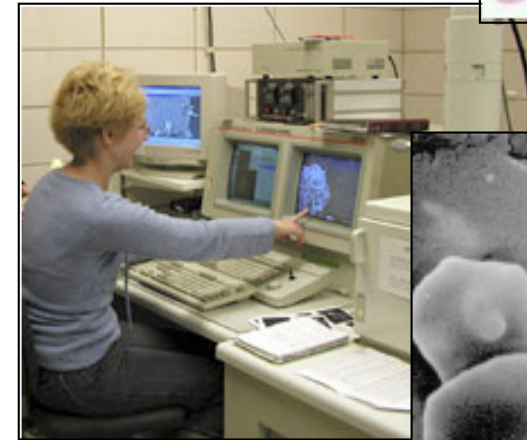
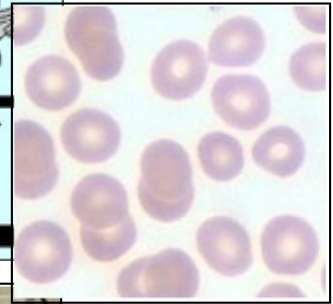
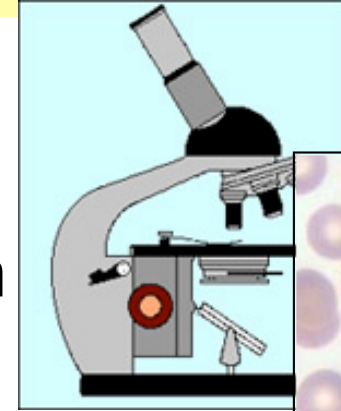
First theoretical introduction by
Richard Feynman
“*There’s Plenty of Room at the Bottom*”
American Physical Society, 1959

- The key is to make use of the unique properties which arise because of the nanoscale.

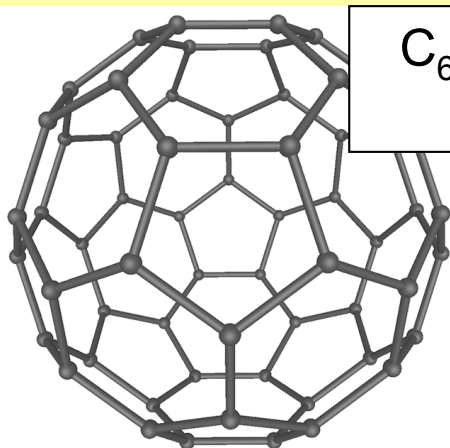


How to See Small Things

- The **naked eye** can see to about $20\mu\text{m}$
- **Light microscopes** let us see to about $1\mu\text{m}$ (bounce **light** off of surfaces to create images)
- **Scanning electron microscopes** (1930s) let us see objects as small as 10 nm (bounce **electrons** off of surfaces to create images)
- **Scanning tunneling microscope** (1980) and **atomic force microscope** (1986) let us see and interact with the atomic structure (use the atom-to-atom interaction to map the surface)



What did we see? New things!

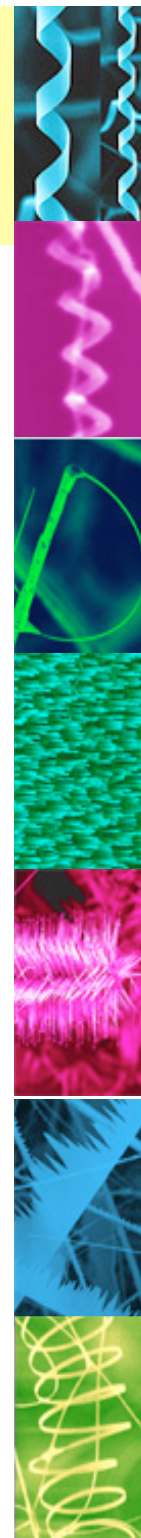
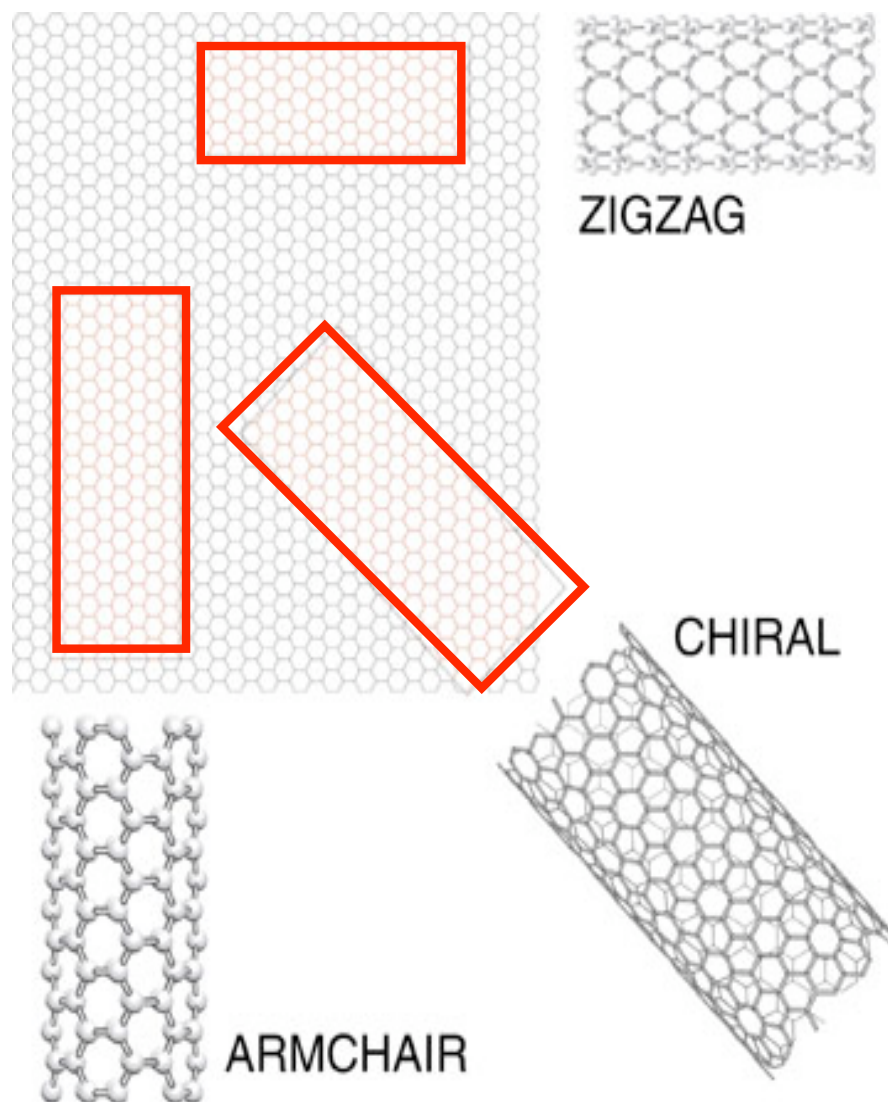


C_{60} Bukyball or Fullerene



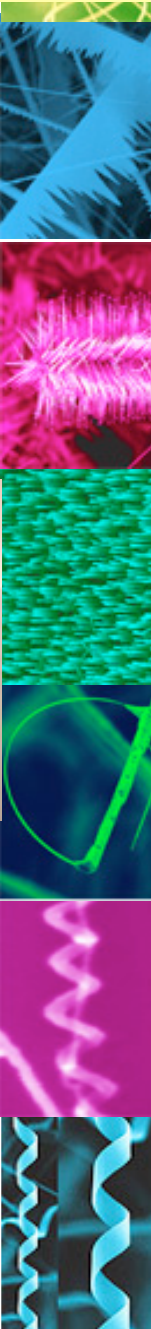
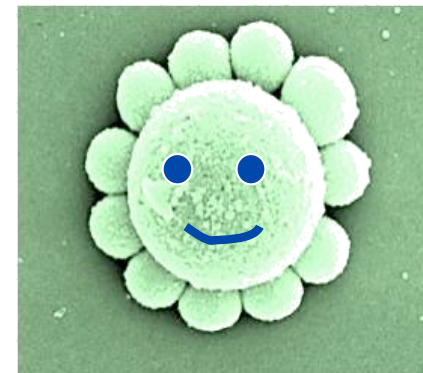
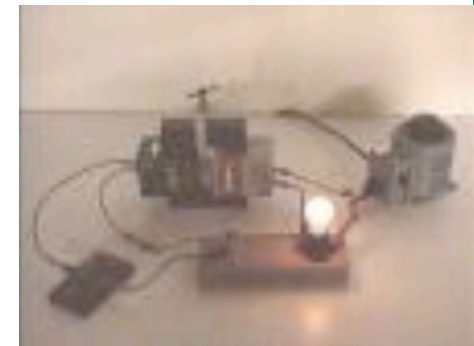
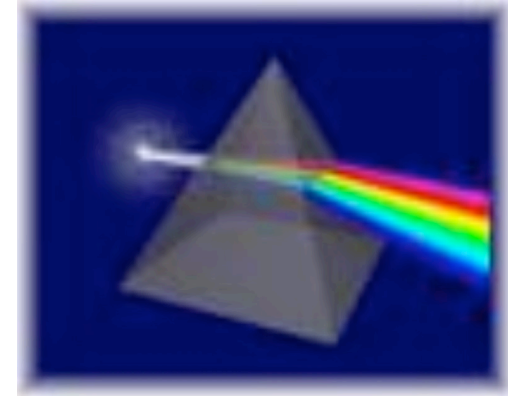
Fuller's American Pavillion,
Monreal Expo '67

Graphene and Nanotubes



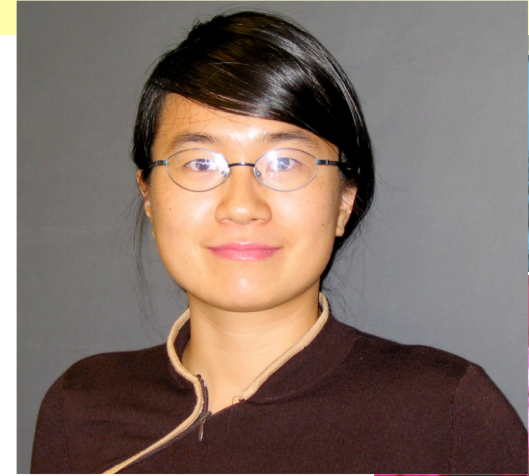
Size-Dependent Properties of Materials

- Types of properties that describe how materials act under certain conditions:
 - Optical (e.g. color, transparency)
 - Electrical (e.g. conductivity)
 - Physical (e.g. hardness, melting point)
 - Chemical (e.g. reactivity, reaction rates)
- At the nanoscale, properties change because different forces dominate at different scales.
- If you are a nanoparticle you are:
 - *Sticky* (intermolecular forces)
 - *Shaky* (thermal energy)
 - *Bumpy* (quantum effects)
 - *Gravity doesn't matter* (really weak)

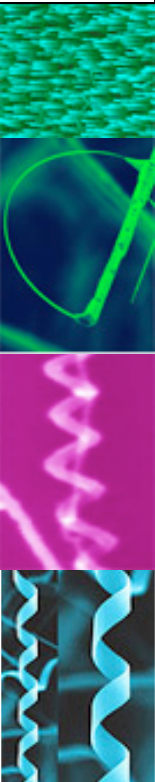
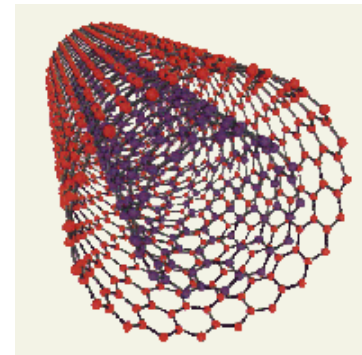
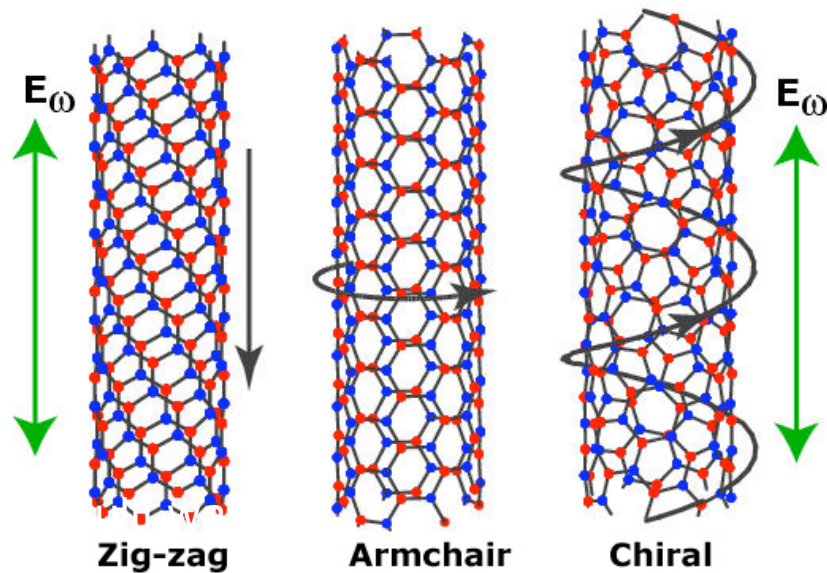


Electrical Properties: Conductivity of Nanotubes

- Because the mass of nanoscale objects is so small, gravity becomes negligible
- Nanotubes are long, thin cylinders of carbon
 - They are 100 times stronger than steel, very flexible, and have unique electrical properties
- Their electrical properties change with diameter, “twist”, and number of walls
 - They can be either conducting or semi-conducting



Jeanie Lau



Energy Applications

- Solid State Lighting (semiconductor)

Filament light bulbs waste about 90% of the input energy in heating.



Robert Haddon



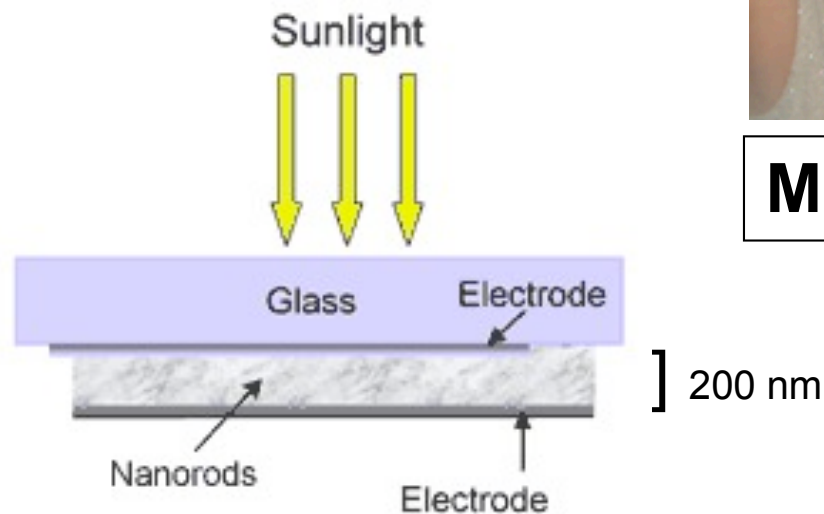
If every light bulb in the USA is replaced with SSL technology, the total national electricity consumption would be reduced of 10%.

[DOE report, 2007]



Environment: Nano Solar Cells

- Solar cells are limited by their efficiency
- Nano solar cells mixed in plastic could be painted on buses, roofs, and clothing
 - Solar becomes a cheaper energy alternative!



Nano solar cell: Inorganic nanorods embedded in semiconducting polymer, sandwiched between two electrodes



Mihri Okzam

